Symmetry Breaking of Novel C₂ Chiral **Across-Ring 1,3-Dienes**

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Received July 6, 1998

Enantiomerically pure C₂-symmetric compounds are powerful tools for chemistry.¹ Nevertheless, their use as chiral templates for the synthesis of unsymmetrical targets is only efficient provided that no additional elements or steps are required to improve the statistical results of the symmetrybreaking step.² Monofunctionalization has been the unique strategy described for C₂ symmetry breaking.^{3,4} Up to now, the possibility of a C_2 symmetry breaking induced by a symmetry-driven functionalization at both homotopic sites remains experimentally unexplored.

We herein report on this possibility by using the formation of the unsymmetrical structures II as model. In this context, the Diels–Alder reaction of the hitherto unknown chiral C_2 across-ring 1,3-dienes $\mathbf{I}^{5,6}$ and a $D_{\infty h}$ symmetric acetylene (Scheme 1) has been carried out. The driving force for C_2 symmetry breaking is now the conservation of the orbital symmetry⁷ of a $4\pi^{s} + 2\pi^{s}$ process. In addition, dienes **I** have been transformed into the structures III by an osmium-

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Scheme 1. C₂ Symmetry-Breaking of Dienes I by Means of a Monofunctionalization Strategy (a) and by a Symmetry-Driven Strategy (b)



^a Key: (i) p-NO₂C₆H₄CO₂H (YOH), PPh₃, DEAD, THF, rt; (ii) LiOH (1 N), THF/MeOH (2:1); (iii) ClCH2OEt (Cl-EOM), i-Pr2EtN, CH2Cl2, -20 °C to rt; (iv) *n*-BuLi, ClSnMe₃, THF, -78 °C to rt; (v) Cu(NO₃)₂-(OH₂)₃, THF, rt; (vi) dimethylhexylsilyl chloride (Cl-THS), imidazole, DMF, 0 °C to rt; (a) i, ii (71%, two steps); (b) iii (93%), iv (83%); (c) v (88%); (d) i (72%); (e) vi (87%), ii (96%); (f) iii (93%), iv, v (61% two steps); (g) vi (92%), iii (95%), iv (78%); (h) v (73%).

9. X = OTHS

7, X = OH, P = Y _

8, X = OTHS, P = H - e

catalyzed bis-hydroxylation reaction. This process⁸ illustrates that monofunctionalization, as defined above, can also be applied on dienes I to induce C_2 symmetry breaking.

Preparation of bis-2,2'-cyclohexenol derivative 3 was first addressed (Scheme 2). Both enantiomers of the chosen starting precursor, 2-bromo-2-cyclohexenol (1), can be efficiently prepared,⁹ and they can be interconnected¹⁰ by taking advantage of their chirality plane.¹¹ Therefore, structures II and III or their enantiomers could be synthesized from 1 or ent-1.

The protection of the allylic alcohols of 1 and *ent*-1 as their ethoxymethoxy derivatives (EOM) followed by transmetalation (n-BuLi, THF, -78 °C) and subsequent treatment with trimethyltin chloride (THF, -78 °C to rt) gave, respectively, the corresponding trimethylvinylstannanes 2 and ent-2 (77% from 1 and ent-1, respectively). The key homocoupling processes from each 2 and ent-2 were performed by treatment with Cu(NO₃)₂(OH₂)₃ (THF, rt, 88%).^{12,13} When the same protocol was used starting from rac-1, a mixture of rac-3 and meso-3 (1:1)¹⁴ was obtained.

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^a Key: (a) DMAD, PhH, reflux, 86% for 10, 97% for 11; (b) (i) (E)-NO₂CH=CHSO₂Ph, reflux, (ii) *n*-Bu₃SnH, AIBN, PhMe, reflux, 70% for 12 (two steps), 73% for 13 (two steps).

The introduction of two additional oxygenated groups to the structure of diene 3 prompted us to prepare dienes 6 and 9 to facilitate eventual synthetic manipulations on the cyclohexenyl moiety. Scheme 2 shows the synthesis of diastereoisomeric dienes 6 and 9 from enantiomerically pure diol 4.15

We subsequently focused on the C_2 symmetry breaking of dienes **3** and **6** induced during $4\pi^{s}-2\pi^{s}$ processes. We used two different 2π partners, dimethyl acetylenedicarboxylate (DMAD) ($D_{\infty h}$ symmetry) and a synthetic equivalent of acetylene¹⁶ (Scheme 3). We have used the sequence Diels-Alder reaction of dienes 3 and 6 with (E)-1-nitro-2-(phenylsulfonyl)ethylene¹⁷ followed by treatment of the crude mixture with tri-*n*-butyltin¹⁸ as synthetic equivalent of acetylene. In all cases, adducts 10-13 (86, 97, 70, and 73% yield, respectively) were formed as unique products. The most important features¹⁹ of the unsymmetrical structure of such products are as follows: (1) a global concave-convex shape, (2) an equatorial orientation of the substituent on C-5 (see Scheme 3 for numbering), and (3) the disposition over the convex surface of the axial substituent on C-4 (Scheme 3).

Remarkably, a complete control of the regiochemistry can be coupled whith the C_2 symmetry breaking when an

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^a Key: (a) 4-methylmorpholine N-oxide, OsO₄, acetone, H₂O, rt, 77% for 16, 63% for 17; (b) t-BuOOH, VO(acac)₂, PhH, rt, 70%.

appropiate dienophile is used. Thus, reaction of **3** with (a) 1-cyanovinyl acetate followed by one-pot reduction with excess NaBH₄ or with (b) fumaronitrile yielded 14 (75%) and 15 (82%), respectively, as unique products (Scheme 3).

On the other hand, we decided to study the C_2 symmetrybreaking process induced by the osmium-catalyzed bishydroxylation reaction of dienes 3 and 6. Under standard conditions,²⁰ (4-methylmorpholine N-oxide, OsO₄, acetone, rt), diols 16 (77%) and 17 (63%) were the only products formed (Scheme 4). In this case, proximity, as defined by Bertz,^{2a} is now active. In addition, functionalization of diols 16 and 17 as allylic alcohols allows further substratedirectable reactions. Thus, as a preliminary example, we have found that the vanadium-catalyzed epoxidation of 16 (t-BuOOH, VO(acac)₂, PhH, rt)²¹ yields epoxy diol 18 (70%) (Scheme 4).

In conclusion, we have described two different strategies for the efficient C_2 symmetry breaking of enantiomerically pure novel chiral C_2 across-ring 1,3-dienes.²² The first one, which is unprecedent, uses a symmetry-driven functionalization. The second one is based on the usual monofunctionalization strategy. A synthetic correlation between the unsymmetrical backbone of the morphinans and of the bicyclic core of zaragozic acids (squalestatins) throughout dienes **I** is currently underway as a further application of these new methods.

Acknowledgment. Financial support by DGICYT (Grant No. PB95-0065) and DGES (Grant No. PB96-0832) is acknowledged. G.G.-R. is grateful to the Ministerio de Educación y Cultura for a predoctoral fellowship. We are grateful to Dr. Gregg Whited (Genencor International, Inc.) for the generous gift of the *cis*-diol for **4** and to Drs. M. Bernabé and B. Herradón for helpful discussions.

Supporting Information Available: Experimental procedures and spectral data for compounds 3, meso-3, 6 and 9-18. Molecular mechanics protocol and discussion. Table 1 showing the steric energy values for the conformers of **10** and **12**. Figures with a view of the most stable conformers (15 pages).

JO981296I

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